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Key indicators

Single-crystal X-ray study
 $T = 296$ K
Mean $\sigma(\text{C}-\text{C}) = 0.002$ Å
 R factor = 0.039
 wR factor = 0.113
Data-to-parameter ratio = 15.9For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.2,3,6,7-Tetrahydrobenzo[1,2-*b*;4,5-*b'*]difuran

In the crystal structure of the title compound, $\text{C}_{10}\text{H}_{10}\text{O}_2$, the molecule has crystallographic inversion symmetry. Each dihydrofuran ring adopts an envelope conformation and, excluding the flap C atom, is essentially coplanar with the benzene ring.

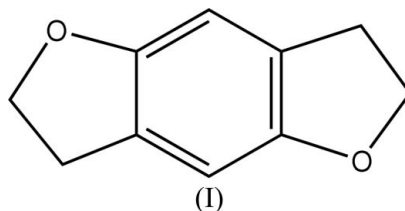
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Comment

Dihydrofurans are an important class of heterocycles with widespread occurrence in nature. Possessing a variety of biological activities, they are used as pharmaceuticals, flavours, insecticides and fish antifeedant agents (Dean, 1982).



The title compound, (I), crystallizes with half a molecule in the asymmetric unit, the molecule lying on an inversion centre. All bond lengths and angles lie within their expected ranges (Allen *et al.*, 1987). Each dihydrofuran ring adopts an envelope conformation, with atom C1 deviating by 0.190 (2) Å from the mean plane through the other four atoms. This mean plane forms a dihedral angle of 0.32 (6)° with the benzene ring.

There are no π - π stacking interactions in the crystal structure of (I). The packing is shown in Fig. 2.

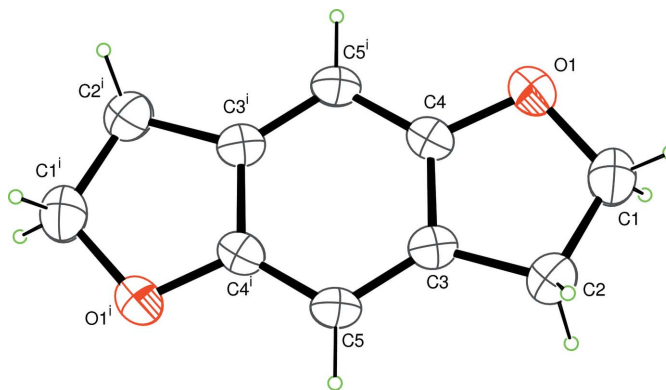


Figure 1

A view of (I). Displacement ellipsoids are drawn at the 30% probability level and H atoms are shown as small circles of arbitrary radii. [Symmetry code: (i) $1 - x, 1 - y, 1 - z$.]

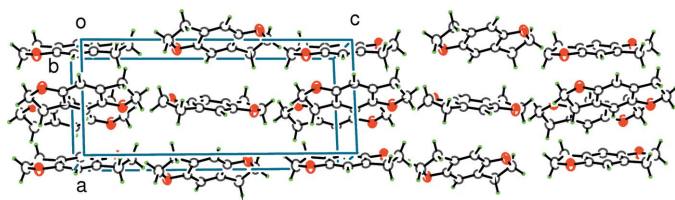


Figure 2
The molecular packing of (I).

Experimental

1,4-Dibromo-2,5-bis(2-chloroethoxy)benzene (3.93 g, 10 mmol) was dissolved in dry tetrahydrofuran (100 ml) in a nitrogen atmosphere and cooled to 273 K. A solution of 2.0 M *n*-butyllithium (5.5 ml, 11 mmol) was added quickly. The mixture was stirred at room temperature for 10 min, then extracted with diethyl ether and water, dried with anhydrous MgSO_4 , and the solid product recrystallized from diethyl ether to obtain the title compound (0.98 g, yield 60%) (Monte *et al.*, 1996). It was further recrystallized from acetone, giving colourless crystals suitable for X-ray diffraction.

Crystal data

$\text{C}_{10}\text{H}_{10}\text{O}_2$
 $M_r = 162.19$
Orthorhombic, *Pbca*
 $a = 7.434$ (4) Å
 $b = 5.972$ (2) Å
 $c = 17.603$ (9) Å
 $V = 781.4$ (6) Å³
 $Z = 4$
 $D_x = 1.378$ Mg m⁻³

Mo $K\alpha$ radiation
Cell parameters from 5940 reflections
 $\theta = 3.4$ – 27.5°
 $\mu = 0.10$ mm⁻¹
 $T = 296$ (1) K
Block, colourless
 $0.36 \times 0.30 \times 0.20$ mm

Data collection

Rigaku R-AXIS RAPID diffractometer
 ω scans
Absorption correction: none
6952 measured reflections
891 independent reflections

656 reflections with $F^2 > 2\sigma(F^2)$
 $R_{\text{int}} = 0.026$
 $\theta_{\text{max}} = 27.5^\circ$
 $h = -9 \rightarrow 9$
 $k = -7 \rightarrow 7$
 $l = -22 \rightarrow 22$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.039$
 $wR(F^2) = 0.113$
 $S = 1.02$
891 reflections
56 parameters
H-atom parameters constrained

$w = 1/[0.0012F_o^2 + \sigma(F_o^2)]/(4F_o^2)$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.24$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.14$ e Å⁻³
Extinction correction: Larson (1970)
Extinction coefficient: 2.7 (4) $\times 10^2$

H atoms were placed in idealized positions ($\text{C}-\text{H} = 0.93$ – 0.97 Å) and refined as riding on their parent atoms, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

Data collection: *PROCESS-AUTO* (Rigaku, 1998); cell refinement: *PROCESS-AUTO*; data reduction: *CrystalStructure* (Rigaku/MSK, 2004); program(s) used to solve structure: *SIR97* (Altomare *et al.*, 1999); program(s) used to refine structure: *CRYSTALS* (Betteridge *et al.*, 2003); molecular graphics: *ORTEP3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *CrystalStructure*.

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